| ntroduc  | jectives:  |  | ᅵ  | Т   | P  |  |
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| ntroduc  | jectives.  |  |  |   |  |  |
|  |  |  |  |   |  |  |
|  |  | to modern biology with an emphasis on evolution of biology as a aware of biological principles. The course will facilitate the students  |  | D-IIIL  | iscipii  | nar  |
| Realize  | e that all forms of life have the same building blocks.  |  |  |   |  |  |
| Convey   | y that without catalysis life would not have existed on earth.   |  |  |   |  |  |
|  |  |  |  |   |  |  |
| Compreworld.   | ehend the f  | undamental principles of energy transactions are the same in physi-  | cal  | and   | biolo  | gica   |
|  | tand the fu  | ndamentals about the molecular basis of coding and decoding  |  |   |  |  |
| I TI   | BIOMOLE  | CULES  | T  | 6   | +  | 3  |
| enclatu  | ire - struc  | ture of enzymes – enzyme cofactors- properties of enzymes(cat  | taly   | rtic p  | rope   | ties   |
| cture or   | ıly; Amino   | acids- classification- amphoteric nature of amino acids - zwitter ion  | - is   | oele  | ctric  | ooir   |
|  |  |  |  |   | 3  |  |
| ificity,   | reversibility  | , sensitiveness to heat and inhibitors, colloidal nature)- mechanisi   | m c  | of th   | e enz  | ym   |
|  |  |  |  |   |  |  |
|  |  | ·  | $\top$   | <b>C</b>  |  | _  |
| 11 111   | WACKOW   | OLECOLES   |  | 0   | Т  | 3  |
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|  |  |  |  |   |  |  |
|  |  | of protein synthesis.  |  |   | ,  |  |
| 10010-1  |  |  | Т  | ^   |  |  |
| IT IV  | METABO   | LISM   |  | 6   | +  | 3  |
| IT IV  | amics as a   | pplied to biological systems - exothermic and endothermic versus   | e e  | nder  | gonic  | an   |
| IT IV<br>modyna<br>gonic re  | amics as a   | applied to biological systems - exothermic and endothermic versus<br>oncept of equilibrium constant and its relation to standard free ener   | gy.  | nder  | gonic<br>ontane  | an<br>eity   |
| IT IV<br>modynagonic recture of<br>hase- k   | amics as a<br>eactions- c<br>ATP; Glyc<br>kinds of rea   | applied to biological systems - exothermic and endothermic versus<br>oncept of equilibrium constant and its relation to standard free ener<br>olysis- definition- flow chart- steps involved in glycolysis- preparator<br>actions in glycolysis; Photosynthesis- definition- significance photosy  | gy.  | nder  | gonic<br>ontande<br>e and  | an<br>eity<br>pa   |
| IT IV<br>modynagonic recture of<br>hase- k   | amics as a<br>eactions- c<br>ATP; Glyc<br>kinds of rea<br>ture of pign   | applied to biological systems - exothermic and endothermic versus<br>oncept of equilibrium constant and its relation to standard free ener<br>olysis- definition- flow chart- steps involved in glycolysis- preparator<br>actions in glycolysis; Photosynthesis- definition- significance photosynents factors affecting photosynthesis- external and internal factors.  | gy.  | nder<br>- spo<br>ohaso<br>etic-   | gonic<br>ontane<br>e and<br>pigm   | an<br>eity<br>pa<br>ent  |
| IT IV<br>modynagonic recture of<br>hase- k<br>s- struc   | amics as a<br>eactions- c<br>ATP; Glyc<br>kinds of rea<br>ture of pign   | applied to biological systems - exothermic and endothermic versus oncept of equilibrium constant and its relation to standard free enerolysis- definition- flow chart- steps involved in glycolysis- preparator actions in glycolysis; Photosynthesis- definition- significance photosynents factors affecting photosynthesis- external and internal factors.  ACIDS   | gy.<br>ry p  | nder<br>- spo<br>bhase<br>etic-   | gonic<br>ontande<br>e and<br>pigm  | an<br>eity<br>pa<br>ent  |
| IT IV modynagonic recture of hase- ks- structure of the second of the se | amics as a eactions- c ATP; Glyc kinds of rea ture of pign NUCLEIC ctural comp   | applied to biological systems - exothermic and endothermic versus<br>oncept of equilibrium constant and its relation to standard free ener<br>olysis- definition- flow chart- steps involved in glycolysis- preparator<br>actions in glycolysis; Photosynthesis- definition- significance photosynents factors affecting photosynthesis- external and internal factors.  | gy-<br>ry p<br>nth   | nderg<br>- spo<br>phase<br>etic-<br>6   | gonic<br>ontane<br>e and<br>pigm<br>+<br>cleosie   | andeity parent   |
|  | Comproworld.  Unders  Unders  UT I  ohydra ture or ions of  IT II  enclatu ifficity, n- lock ion( te mulatic IT III  eins- certies certies certies | Comprehend the forworld.  Understand the further blowning and the furth | Understand the fundamentals about the molecular basis of coding and decoding  IT I BIOMOLECULES  ohydrates- classification - Glucose properties and structural elucidation —fructose, sture only; Amino acids- classification- amphoteric nature of amino acids - zwitter ion ions of amino acids; Vitamins - general characteristics- classification- function and definition and definition in the structure of enzymes — enzyme cofactors- properties of enzymes(catificity, reversibility, sensitiveness to heat and inhibitors, colloidal nature)- mechanism in lock and key mechanism and koshland induced fit mechanism -Factors affecting ion( temperature, pH, substrate concentration, enzyme concentration, water inhibit mulation)- enzyme kinetics —michaelis-menten equation.  IT III MACROMOLECULES  eins- classification- structure of proteins- primary, secondary, tertiary and quaterities of proteins- physical and chemical properties- colour reaction of proteins (biuret | Comprehend the fundamental principles of energy transactions are the same in physical world.  Understand the fundamentals about the molecular basis of coding and decoding  IT I BIOMOLECULES  Ohydrates- classification - Glucose properties and structural elucidation –fructose, sucture only; Amino acids- classification- amphoteric nature of amino acids - zwitter ion - is ions of amino acids; Vitamins - general characteristics- classification- function and deficiental transport of the properties of enzymes and inhibitors, colloidal nature) - mechanism of the control of the presentation of the properties of enzyme concentration, water inhibitors mulation) - enzyme kinetics –michaelis-menten equation.  IT III MACROMOLECULES  Pars - classification - structure of proteins - primary, secondary, tertiary and quaternerities of proteins - physical and chemical properties - colour reaction of proteins (biuret reaction) | Comprehend the fundamental principles of energy transactions are the same in physical and world.  Understand the fundamentals about the molecular basis of coding and decoding  IT I BIOMOLECULES 6  ohydrates- classification - Glucose properties and structural elucidation —fructose, sucrose ture only; Amino acids- classification- amphoteric nature of amino acids - zwitter ion - isoelections of amino acids; Vitamins - general characteristics- classification- function and deficiency of the composition of t | Comprehend the fundamental principles of energy transactions are the same in physical and biologology world.  Understand the fundamentals about the molecular basis of coding and decoding  IT I BIOMOLECULES  6 +  ohydrates- classification - Glucose properties and structural elucidation –fructose, sucrose, starture only; Amino acids- classification- amphoteric nature of amino acids - zwitter ion - isoelectric prions of amino acids; Vitamins - general characteristics- classification- function and deficiency diseases.  IT II ENZYMES  6 +  enclature - structure of enzymes – enzyme cofactors- properties of enzymes(catalytic proper ifficity, reversibility, sensitiveness to heat and inhibitors, colloidal nature)- mechanism of the enzyn- lock and key mechanism and koshland induced fit mechanism -Factors affecting rate of enzymentation (temperature, pH, substrate concentration, enzyme concentration, water inhibitors, end promulation)- enzyme kinetics –michaelis-menten equation. |

| Upon c | ompletion of this course, the students will be able to:  |  |  |  |  |  |
|--------|--|--|--|--|--|--|
| CO1    | . Appreciate that all types of life have the identical structural units.                           |  |  |  |  |  |
| CO2    | Highlight the idea that without catalysis, living beings would not have existed on earth.          |  |  |  |  |  |
| CO3    | Be familiar with the investigation of biological processes at the reduction level.                 |  |  |  |  |  |
| CO4    | : Figure out that the primary principles of energy transactions are alike in physical and chemical |  |  |  |  |  |
| Text B | ooks:  |  |  |  |  |  |
|        | FJ.L.Jain, Sanjay jain and Nitin jain- "Fundamentals of Biochemistry" - Sixth edition, S.Chand and |  |  |  |  |  |
| 1.     | company Ltd., Ram nagar, 2005.   |  |  |  |  |  |
|        | Dr.A.V.S.S.Rama Rao-" Text book of Biochemistry"- Text book of Biochemistry- First edition- UBS    |  |  |  |  |  |
| 2.     | Publishers' Distributors Pvt. Ltd., 2008.  |  |  |  |  |  |
| 3.     | U. Satyanarayana –" Biochemistry"-5th edition – Sri Padmavathi Publications Ltd.,2017.             |  |  |  |  |  |
| Refere | Reference Books:   |  |  |  |  |  |
| 1.     | Campbell, N. A.; Reece, J. B.; Urry, Lisa; Cain, M,L.; Wasserman, S. A.; Minorsky,                 |  |  |  |  |  |
|        | P. V.; Jackson, R. B" Biology: A global approach"- Pearson Education Ltd                           |  |  |  |  |  |
| 2.     | Conn, E.E; Stumpf, P.K; Bruening, G; Doi, R.H-" Outlines of Biochemistry"- John Wiley and Sons     |  |  |  |  |  |
|        | By Nelson, D. L.; and Cox- "Principles of Biochemistry"- V Edition- M. M.W.H. Freeman and          |  |  |  |  |  |
| 3.     | Company  |  |  |  |  |  |
| 4      | Stent, G. S.; and Calender-" Molecular Genetics"- Second edition - R. W.H. Freeman and             |  |  |  |  |  |
| 4.     | company, Distributed by Satish Kumar Jain for CBS Publisher  |  |  |  |  |  |